



The inhibitory effect of manganese on acetylcholinesterase activity enhances oxidative stress and neuroinflammation in the rat brain

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ABSTRACT

Background: Manganese (Mn) is a naturally occurring element and an essential nutrient for humans and animals. However, exposure to high levels of Mn may cause neurotoxic effects. The pathological mechanisms associated with Mn neurotoxicity are poorly understood, but several reports have established it is mediated, at least in part, by oxidative stress.

Objectives: The present study was undertaken to test the hypothesis that a decrease in acetylcholinesterase (AChE) activity mediates Mn-induced neurotoxicity.

Methods: Groups of 6 rats received 4 or 8 intraperitoneal (i.p.) injections of 25 mg MnCl₂/kg/day, every 48 h. Twenty-four hours after the last injection, brain AChE activity and the levels of F₂-isoprostanes (F₂-IsoPs) and F₄-neuroprostanes (F₄-NPs) (biomarkers of oxidative stress), as well as prostaglandin E₂ (PGE₂) (biomarker of neuroinflammation) were analyzed.

Results: The results showed that after either 4 or 8 Mn doses, brain AChE activity was significantly decreased ($p < 0.05$), to $60 \pm 16\%$ and $55 \pm 13\%$ of control levels, respectively. Both treated groups exhibited clear signs of neurobehavioral toxicity, characterized by a significant ($p < 0.001$) decrease in ambulation and rearings in open-field. Furthermore, Mn treatment caused a significant increase ($p < 0.05$) in brain F₂-IsoPs and PGE₂ levels, but only after 8 doses. In rats treated with 4 Mn doses, a significant increase ($p < 0.05$) in brain F₄-NPs levels was found. To evaluate cellular responses to oxidative stress, we assessed brain nuclear factor-erythroid 2 p45-related factor 2 (Nrf2) and Mn-superoxide dismutase (Mn-SOD, SOD2) protein expression levels. A significant increase in Mn-SOD protein expression ($p < 0.05$) and a trend towards increased Nrf2 protein expression was noted in rat brains after 4 Mn doses vs. the control group, but the expression of these proteins was decreased after 8 Mn doses. Taken together, these results suggest that the inhibitory effect of Mn on AChE activity promotes increased levels of neuronal oxidative stress and neuroinflammatory biomarkers.

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1. Introduction

Manganese (Mn) is a transition metal that is essential for normal cell growth and development (Au et al., 2008). Mn acts as a cofactor for a variety of metalloenzymes, such as the mitochondrial protein superoxide-dismutase (Mn-SOD), a critical enzyme in attenuating oxidative stress (Stallings et al., 1991), arginase, which is responsible for urea production in the liver, pyruvate carboxylase, an essential enzyme in gluconeogenesis (Crossgrove and Zheng, 2004), as well as glutamine synthetase, an astrocyte-specific enzyme that converts glutamate into glutamine (Aschner and Gannon, 1994; Burton and Guilarte, 2009). Additional

Mn-dependent enzymes include oxidoreductases, transferases, hydrolases, lyases, isomerases and ligases (Aschner and Aschner, 2005). Clinical manifestations of Mn deficiency include seizures, retarded growth, skeletal abnormalities and impaired reproductive function (Roth and Garrick, 2003) and Mn deficiency during development is also associated with convulsive disorders (Aschner and Gannon, 1994).

At the other end of the spectrum, excessive exposure to Mn may cause neurotoxicity (McMillan, 1999; Aschner et al., 2005) characterized by a frequently irreversible Parkinsonian-like syndrome, which includes fixed gaze, bradykinesia, postural difficulties, rigidity, tremor, dystonia and decreased mental status (Levy and Nassetta, 2003). Neurotoxic effects resulting from excessive Mn exposure were first described by Couper in 1837 in Scottish laborers grinding Mn black oxide in a chemical industry (Couper, 1837; Meyer-Baron et al., 2009). Presently, occupational exposure occurs

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in miners, welders and workers in the ferromanganese-alloy industry and the manufacturing of dry cell batteries (Takeda, 2003; Jankovic, 2005; Burton and Guilarte, 2009). Mn neurotoxicity may also result from parenteral nutrition therapy, especially in patients with liver disease or immature hepatic function, such as the premature neonate (Erikson et al., 2007). In contrast to numerous reports describing Mn toxicity following inhalation exposure in humans, there are relatively few reports on manganism arising from water or dietary sources (Dorman et al., 2001). Kawamura et al. (1941) documented outbreaks of manganism in Japan and Greece, respectively, due to consumption of well-water contaminated with extremely high levels of Mn (1.8–4 mg Mn/l). Wasserman et al. (2006, 2011) also reported adverse impact of Mn exposure associated with water consumption on child developmental outcomes in Bangladesh. In another cross-sectional study, neurological symptoms and elevated hair Mn levels were reported in elderly individuals residing in an area in Greece for more than 10 years who consumed drinking water with Mn levels of 1.8–2.3 mg/l (Kondakis et al., 1989). Toxicological oral studies have been performed in rodents, also demonstrating biochemical changes in the brain following administration of 1 mg MnCl₂·4H₂O/ml in drinking water (approximately 40 mg Mn/kg-day) for over two years (Lai et al., 1981, 1982).

Manganism is linked to increased brain Mn levels primarily in brain regions known also to be iron (Fe)-rich, namely, caudate, putamen, globus pallidus, substantia nigra and subthalamic nuclei. These regions are collectively referred to as the basal ganglia (Erikson and Aschner, 2003). Notably, cholinergic neurons projecting to the striatum synapse with nigral-dopaminergic (DAergic) and gamma-aminobutyric acid (GABA) GABAergic neurons in the striatum (Moore and Wuerthele, 1979).

Studies in humans and animals indicate that excessive Mn exposure is associated with behavioral changes and altered cognitive function (Rohling and Demakis, 2007; Bouchard et al., 2008). Mn-induced cortical cholinergic dysfunction is compatible with these cognitive deficits, as well as with the dementia observed later on in the clinical course of manganism (Finkelstein et al., 2007). Significant inhibition of acetylcholinesterase (AChE) activity was observed following lengthy periods of exposure to Mn in adult (Sitaramayya et al., 1974; Martinez and Bonilla, 1981) and neonatal (Deskin et al., 1981; Lai et al., 1984) rat brain. The inhibition of AChE prevents the hydrolysis of acetylcholine (ACh), leading to accumulation of ACh in the synaptic cleft and overstimulation of muscarinic and nicotinic ACh receptors. Decreased activity of AChE, the enzyme responsible for ACh hydrolysis, was shown to be associated with increased oxidative and nitrosative stress (Milatovic et al., 2006), corroborating other reports on the ability of Mn to increase oxidative stress (Donaldson et al., 1981; Donaldson, 1984; Donaldson and Labella, 1984; Ali et al., 1995; Hamai and Bondy, 2004).

In several experimental pathological conditions, the increased AChE activity is also associated with oxidative stress. Thomé et al. (2011) reported that AChE activity and lipid peroxidation were increased in rats' brain in an experimental model of exposure to aged and diluted sidestream smoke (ADSS). Rosemberg et al. (2010) also reported that ethanol exposure increased AChE activity and lipoperoxidation in zebrafish brain. A study by Schallreuter and Elwary (2007) showed that epidermal AChE is a target to H₂O₂-mediated oxidation of methionine and tryptophan residues. Enzyme kinetics revealed concentration dependent activation/deactivation of AChE by H₂O₂.

Excessive ROS production induces the oxidation of membrane polyunsaturated fatty acids, yielding a multitude of lipid peroxidation products (Halliwell and Gutteridge, 1990). Two families of such products are the F₂-isoprostanes (F₂-IsoPs) and F₄-neuroprostanes (F₄-NPs). F₂-IsoPs are a group of arachidonic acid (AA)-derived prostanoid isomers generated from the free radical-initiated

peroxidation of AA independent of cyclooxygenase (COX) (Musiek and Morrow, 2005; Milatovic and Aschner, 2009; Roberts and Milne, 2009). F₂-IsoPs are isomeric to COX-derived prostaglandin (PG) F_{2α} (Greco and Minghetti, 2004). Analogously, auto-oxidation of the omega-3 polyunsaturated fatty acid docosahexaenoic acid (DHA) generates an array of IsoP-like compounds termed neuroprostanes (NPs) (Musiek et al., 2005). Unlike the fatty acid AA, DHA is highly concentrated in neuronal membranes, reflecting F₄-NPs oxidative damage to neuronal membranes (Roberts et al., 1998; Reich et al., 2000, 2001). Therefore, the measurement of F₄-NPs provides a more selective index of neuronal oxidative damage, while the measurement of F₂-IsoPs provides an index of global brain oxidative damage, integrating stress levels both in glial and neuronal cells (Arneson and Roberts, 2007). ROS generation is also associated with inflammatory responses and release of inflammatory mediators including prostaglandins, which are derivatives of AA oxidation catalysed by the oxidation of COX-2 (Milatovic et al., 2009). Herein, we assessed the effect of Mn on neuroinflammation by measuring prostaglandin E₂ (PGE₂) levels.

NF-E2-related factor 2 (Nrf2) is a leucine zipper transcription factor belonging to the cap 'n' collar family; it is involved in the induction of genes encoding antioxidant proteins, including those in the GSH family. Given the ability of Mn to induce oxidative stress, we determined its protein expression levels in response to Mn treatment. We also analyzed the expression of Mn-SOD, a mitochondria-specific anti-oxidant enzyme, given that Mn preferentially accumulates in this organelle, where it disrupts oxidative phosphorylation and increases ROS generation (Gunter et al., 2006).

Given the above, the present study was conducted to investigate the hypothesis that the *in vivo* modulatory effect of Mn on AChE activity is associated with oxidative damage, neuroinflammation and the up regulation of oxidative stress defense enzymes (Mn-SOD) and transcription factors (Nrf2).

2. Experimental procedure

2.1. Materials

Manganese chloride tetrahydrate (MnCl₂·4H₂O; 99.99%), Mn standard for Graphite Furnace Atomic Absorption Spectrometry (GFAAS), acetylthiocholine (ATCh) iodide, hydrogen peroxide (H₂O₂, 30%) and 5,5'-dithiobis(2-nitrobenzoic acid) (DTNB) were purchased from Sigma-Aldrich (St. Louis, MO). 15-F_{2α}-IsoP-d₄ (internal standard for F₂-IsoPs that contains four deuterium atoms), PGE₂ internal standard and prostaglandin F₂, E₂ and D₂ methyl esters were purchased from Cayman Chemicals (Ann Arbor, MI). Nitric acid (HNO₃; 65%) and magnesium matrix modifier for GFAAS were purchased from Merck (Darmstadt, Germany). Albumin standard was purchased from Pierce (Rockford, USA). Rabbit polyclonal antibody Nrf2 (sc-722) was purchased from Santa Cruz Biotechnology (CA, USA). Rabbit polyclonal antibody Mn-SOD was purchased from Stressgen Biotechnologies Corporation (San Diego, CA). Monoclonal anti-β-actin antibody produced in mouse was purchased from Sigma-Aldrich (St. Louis, MO). Goat primary antibody to rabbit IgG horseradish peroxidase (HRP) was purchased from Abcam (San Francisco, CA, USA) and goat primary anti-mouse IgG HRP labeled secondary antibody was purchased from KPL (MD, USA). The chemiluminescent kit was purchased by Pierce Biotechnology. The HyBlot CL autoradiography film was purchased from Denville Scientific Inc., Metuchen, NJ. Bio-Rad western-blot assay reagents were used (Bio-Rad Laboratories, Hercules, CA).

2.2. Animals and treatment

Six-week-old male Wistar rats (180 ± 15 g), Specific Pathogen Free, were obtained from Charles River Laboratories (Barcelona, Spain) and maintained under standard environmental conditions. The animals were housed in a room, with 12–12 h light/dark cycle, 50–70% humidity at 24 °C. All experiments were performed in accordance with the Guiding Principles for the University's Care and Use of Animals, University of Lisbon, Portugal. The rats were randomly divided into 3 groups as follows: (A) control (saline solution-treated) with 4 and 8 doses (n = 6 each), (B) Mn 4 doses (25 mg/kg bodyweight, as MnCl₂·4H₂O, n = 6), and (C) Mn 8 doses (25 mg/kg bodyweight, as MnCl₂·4H₂O, n = 6).

All rats received intraperitoneal (i.p.) injections on alternate days. The schedule of Mn injections over time permits the accumulation of detectable brain Mn levels, while lessening the effects of acute exposure in other organs. This methodology

is based on the long half-life of Mn in the brain (51–74 days) relative to its short half-life in visceral organs such as the heart and the liver (Silva and Bock, 2008).

Mn speciation is another important factor determinant for toxicity. Mn²⁺ is the predominant form in biological systems. The valence state of Mn may also influence its disposition in the body. Roels et al. (1997) showed that more soluble forms of Mn (e.g., MnCl₂ vs. MnO₂) given to three-month-old rats given via intratracheal instillation are more readily delivered to the brain.

The animals were euthanized with pentobarbital (20 mg/kg), 24 h after the last dose followed by cervical dislocation. The cerebral hemispheres were immediately removed and stored at –80 °C until biochemical analyses and chemical determinations.

2.3. Behavioral assays

Locomotor activity was evaluated in an open-field apparatus. Animals were individually placed at the center of the open-field apparatus (60 cm × 90 cm × 30 cm, divided into six equal squares). Spontaneous ambulation (number of segments crossed with the four paws) and rearing (exploratory activity expressed by the number of rearings on the hind limbs) were recorded for 5 min (Markel et al., 1989; Marreilha dos Santos et al., 2011).

2.4. Determination of brain AChE activity

AChE (EC 3.1.1.7) activity was determined by the method of Ellman et al. (1961), which is based on the formation of the yellow 5-thio-2-nitrobenzoate anion produced in the reaction between 5,5'-dithiobis-(2-nitrobenzoic acid) (DTNB) and thiocholine, after the AChE-mediated hydrolysis of ATCh. Samples were weighed (100 mg of the brain homogenate) and 1.8 ml of 0.1 M buffer sodium phosphate dibasic anhydrous/sodium phosphate monobasic pH 7.4 and 0.2 ml of 5% Triton were added to each tube. Briefly, the reaction was initiated after the addition of 0.05 ml of DTNB and 0.05 ml of ATCh iodide, which was used as substrate. The final concentrations of DTNB and substrate were 0.33 mM and 1.56 mM, respectively. The reaction was followed spectrophotometrically by the increase of absorbance at 412 nm. The activity was expressed as nanomole of ATCh hydrolyzed per minute per milligram protein. An aliquot of the homogenate was used for protein determination. Total protein was measured with the method of Bradford (1976). Bovine serum albumin was used as the standard. While the DTNB method has some limitations and is not fully specific, it has been widely used in assays analogous to those described herein (Cederblad et al., 1986).

2.5. Quantification of PGE₂ in brain

PGE₂ was measured by a stable isotope dilution gas chromatographic/negative ion chemical ionization–mass spectrometric assay. [⁴H₂]-PGE₂ (2 ng) and [⁴H₂]-PGD₂ (4.25 ng) were added to brain homogenates after extraction with an indomethacin 10^{–6} M methanolic solution. The sample was then acidified to pH 3 and extracted on a C18 Sep-Pak cartridge. PGE₂ was eluted with a mixture of ethyl acetate:heptane and evaporated under a stream of N₂. PGE₂ in methoxyamine solution was extracted with ethyl acetate and dried under a N₂ stream. The pentafluorobenzyl esters were formed by incubation with pentafluorobenzyl bromide (10% in acetonitrile) and diisopropylethanolamine. The pentafluorobenzyl esters were purified by thin layer chromatography (TLC), converted to O-methylloxime pentafluorobenzyl ester trimethylsilyl derivatives after incubation with N,O-bis(trimethylsilyl)-trifluoroacetamide and dimethylformamide and finally, PGE₂ was dissolved in undecane that dried over a bed of calcium hydride. Gas chromatographic/negative ion chemical ionization–mass spectrometric analysis was performed as described previously with the M-181 ions for PGE₂ (*m/z* 526) and the [⁴H₂]-PGE₂ as internal standard (*m/z* 528) (Milatovic et al., 2009).

2.6. Quantification of F₂-IsoPs and F₄-NPs in brain

Total F₂-IsoPs and F₄-NPs were determined with a stable isotope dilution method with detection by gas chromatography (GC)/mass spectrometry (MS) and selective ion monitoring. Briefly, brain samples were homogenized in Folch solution and the phospholipids were first extracted from the tissue sample and submitted to alkaline hydrolysis to release the F₂-IsoPs and F₄-NPs esterified in membranes. The brain homogenates were acidified to pH 3, followed by the addition of 0.1 ng of 15-F_{2α}-IsoP-d₄ internal standard. F₂-IsoPs and F₄-NPs were subsequently purified by C18 and silica Sep-Pak extraction and by thin layer chromatography (TLC). For the quantification of F₄-NPs instead of scraping 1 cm below and 1 cm above where PGF_{2α} methyl ester migrates on TLC for analysis of F₂-IsoPs, the area scraped was extended to 3 cm above where PGF_{2α} methyl ester migrates. F₂-IsoPs and F₄-NPs were then converted to a trimethylsilyl ether derivative, and quantified by negative ion chemical ionization mass spectrometry. For quantification purposes of F₂-IsoPs we compared the integrated peak area of derivatized F₂-IsoPs (*m/z* 569) with the area of the deuterated internal standard peak (*m/z* 573). Quantification of F₄-NPs levels was achieved through single ion monitoring (SIM) and comparing the area of the appropriate peaks in the *m/z* 593 SIM chromatogram of the F₄-NPs to that of the single peak of the internal standard in the *m/z* 593 SIM chromatogram (Roberts

et al., 1998; Musiek and Morrow, 2005; Arneson and Roberts, 2007; Milatovic and Aschner, 2009).

2.7. Western blot analysis

Rat brain homogenates were prepared in radioimmunoprecipitation assay (RIPA) buffer containing protease inhibitors [4-(2-aminoethyl)benzenesulfonyl fluoride, pepstatinA, E-64, bestatin, leupeptin and aprotinin]. All of the samples were heated for 5 min at 95 °C and centrifuged at 13,000 × *g* for 0.5 min to remove insoluble materials. Equal amounts of protein were loaded in the SDS-PAGE gel and the protein concentrations were determined with the Bradford method (1976). After electrophoresis, the protein was transferred onto a nitrocellulose membrane blocked with 5% non-fat milk, and incubated overnight with a primary antibody. Nrf2 primary antibody (0.1 mg/ml) and Mn-SOD primary antibody (2.5 × 10^{–5} mg/ml) were applied overnight at room temperature, after which immunoreactivity was detected with HRP-conjugated secondary antibody. Protein–antibody complexes were detected with an enhanced chemiluminescent kit (Pierce Biotechnology, Rockford, IL) and exposed to HyBlot CL autoradiography film (Denville Scientific Inc., Metuchen, NJ). Molecular weights of Nrf2, Mn-SOD and β-actin proteins were approximately 68, 25, and 42 kDa, respectively. Individual blot densities were quantified by AlphaEaseFC 4.0. The individual blot densities of Mn-SOD and Nrf2 proteins were normalized to the loading control β-actin.

2.8. Brain Mn analysis

Brain Mn concentrations were determined by GFAAS, using a PerkinElmer Analyst™ 700 atomic absorption spectrometer equipped with an HGA Graphite Furnace and a programmable sample dispenser (AS 800 Auto Sampler) and WinLab 32 for atomic absorption software. Prior to GFAAS analysis, brain samples were digested with an oxidizing acid mixture of 4:1 (v/v) 65% suprapure nitric acid:hydrogen peroxide, using a microwave-assisted acid digestion. Mg (NO₃)₂·6H₂O (0.84 mol/l) was used as a chemical modifier.

2.9. Statistical analysis

The data were analyzed by one-way analysis of variance (ANOVA) followed by Bonferroni's multiple comparison test, except for the evaluation of the changes in bodyweight, where the statistical analysis of significance was carried by unpaired *t*-test. All analyses were carried out with GraphPad Prism 4.02 software for Windows (GraphPad Software, San Diego, CA, USA). Results were considered statistically significant at values of *p* < 0.05. The AlphaEaseFC 4.0 software was used to calculate optical density, and Prism 4.0 (GraphPad, San Diego, CA, USA) was used to calculate and plot the results.

3. Results

3.1. Bodyweight

Bodyweight was assessed at different time-points during the experimental period. Fig. 1 shows the rats' bodyweights at three different time-points: pre-exposure (immediately before the first Mn dose), during exposure (the middle of the study) and post-exposure (at the end of the study). As shown in Fig. 1, a decrease in bodyweight gain was inherent to the Mn-treated groups (either 4 or 8 doses of MnCl₂ 25 mg/kg every 48 h) compared to controls, but the effect was statistically indistinguishable from the controls (*p* < 0.05).

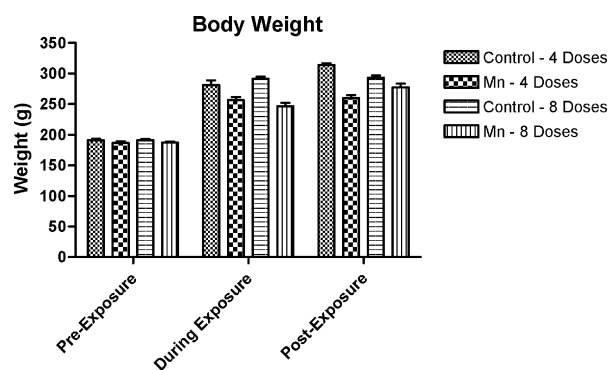


Fig. 1. Changes in rats' bodyweights after intra-peritoneal injection of 4 or 8 doses of MnCl₂ 25 mg/kg and the control group. Data are expressed as mean ± SEM (*n* = 6).

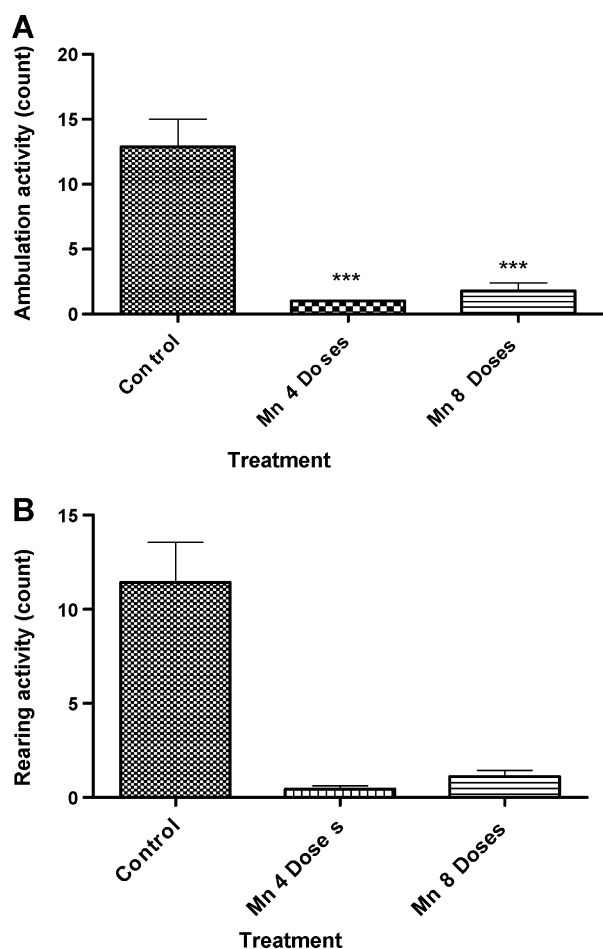


Fig. 2. Neurobehavioral evaluation in rats intra-peritoneally injected with 4 or 8 doses of MnCl_2 25 mg/kg and the control group; (A) ambulation, number of crossings in the open field and (B) number of rearings in the open field. Bars represent mean \pm SEM ($n=6$). *** indicates statistical difference from the control group by one-way ANOVA followed by Bonferroni's multiple comparison tests ($p < 0.001$).

3.2. Locomotor activity assessment

The locomotor activity was assessed by the open-field test. Two parameters, ambulation (horizontal activity) and rearing (vertical activity) were analyzed at two different time points, before the beginning of the experiment and immediately prior to the animals'

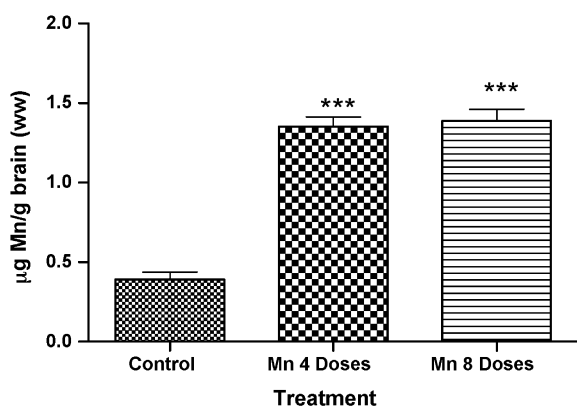


Fig. 3. Brain Mn concentrations in rats intra-peritoneally injected with 4 or 8 doses of MnCl_2 25 mg/kg and the control group. Bars represent mean \pm SEM ($n=6$). *** indicates statistical difference from control group by one-way ANOVA followed by Bonferroni's multiple comparison tests ($p < 0.001$).

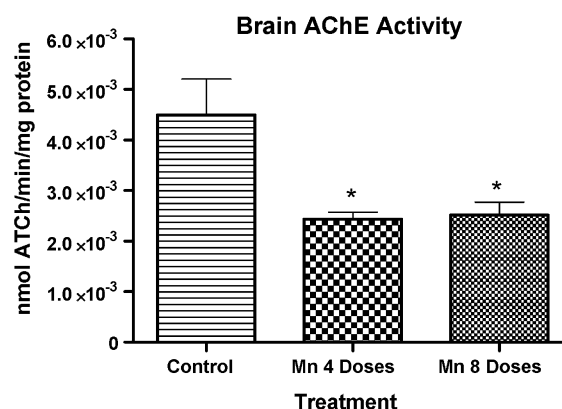


Fig. 4. Effect of Mn on brain AChE activity in rats intra-peritoneally injected with 4 or 8 doses of MnCl_2 25 mg/kg and the control group. Values of AChE represent mean \pm SEM ($n=6$). * significant difference between control and Mn-treated rats ($p < 0.05$) by one-way ANOVA followed by Bonferroni's multiple comparison tests.

sacrifice. Fig. 2 illustrates the effect of Mn treatment on locomotor activity. Mn treatment caused a significant decrease ($p < 0.001$) in ambulation (Fig. 2A) and the number of rearings (Fig. 2B) in the open-field (at both Mn doses) compared with the control group. Neither of these parameters was statistically different among the 4 and 8 dose Mn treated groups.

3.3. Brain Mn concentrations

Brain Mn concentrations are shown in Fig. 3. Significantly higher ($p < 0.001$) brain Mn concentrations were observed both in the 4 and 8 dose Mn treated groups compared with the control group. Mn accumulation in the brain did not increase in a dose-dependent manner and Mn concentrations in the 4 dose Mn treatment group were indistinguishable from those in the 8 dose Mn treatment group.

3.4. Brain AChE activity

Four doses of Mn caused a significant decrease ($p < 0.05$) in brain AChE activity 24 h after the last injection (Fig. 4). Brain AChE activity was statistically indistinguishable from the two Mn treatment groups (4 or 8 doses).

3.5. F_2 -IsoPs and F_4 -NPs levels in brain

We tested the ability of Mn to induce oxidative stress in rats' brains by measuring F_2 -IsoPs and F_4 -NeuroPs, lipid peroxidation biomarkers of oxidative injury. Fig. 5 shows a trend towards an increase in F_2 -IsoPs levels in the 4 dose Mn treated group, but the effect was statistically indistinguishable from the control group. In contrast, treatment with 8 Mn doses caused a significant increase ($p < 0.001$) in F_2 -IsoPs levels compared with the control and the 4 doses Mn treated group ($p < 0.05$).

F_4 -NPs levels in the brain of rats treated with either 4 or 8 Mn doses caused an increase in F_4 -NPs levels vs. the control group. However, this effect was only statistically significant ($p < 0.05$) for the 4 dose Mn treated group (Fig. 6).

3.6. PGE_2 levels in brain

As shown in Fig. 7, Mn induced a dose- and time-dependent increase in PGE_2 levels, but the effect was statistically significant ($p < 0.01$) only in the 8 dose Mn treated group vs. the control group.

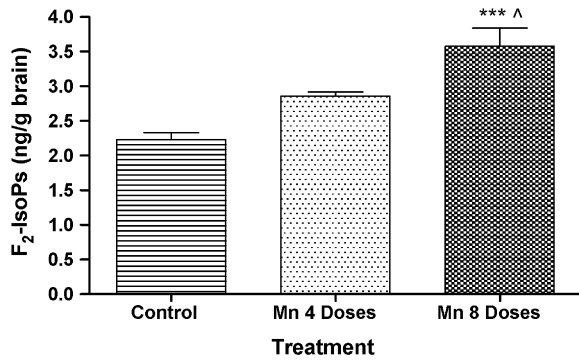


Fig. 5. Cerebral F₂-IsoPs concentrations in rats intra-peritoneally injected with 4 or 8 doses of MnCl₂ 25 mg/kg and the control group. Values of F₂-IsoPs represent mean \pm SEM ($n=6$). *** $p < 0.01$ significantly different from control group, $\wedge p < 0.05$ significantly different from 4 doses Mn group, by one-way ANOVA followed by Bonferroni's multiple comparison tests.

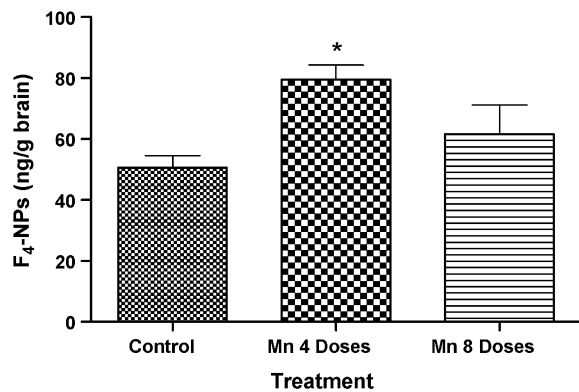


Fig. 6. Cerebral F₄-NPs concentrations in rats intra-peritoneally injected with 4 or 8 doses of MnCl₂ 25 mg/kg and the control group. Values of F₄-NPs represent mean \pm SEM ($n=6$). * $p < 0.05$ vs. control by one-way ANOVA followed by Bonferroni's multiple comparison tests.

3.7. Mn-SOD and Nrf2 expression in brain

Mn significantly enhanced ($p < 0.05$) brain Mn-SOD protein expression after 4 doses compared with the control group (Fig. 8B); after 8 doses the Mn-SOD levels were statistically indistinguishable from the control group. Either 4 or 8 dose Mn treatment did not result in statistically significant change in brain Nrf2 protein expression compared with the control group (Fig. 8A).

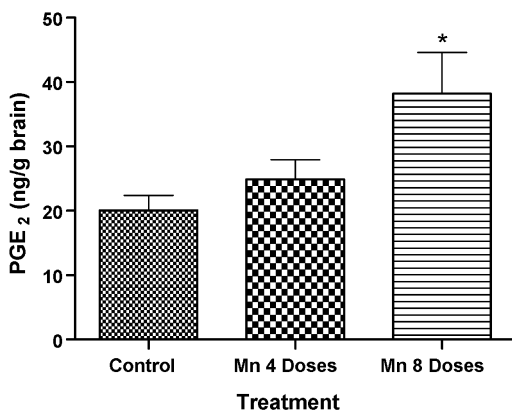


Fig. 7. PGE₂ levels in brain of rats intra-peritoneally injected with 4 or 8 doses of MnCl₂ 25 mg/kg and a control group. Values of PGE₂ represent mean \pm SEM ($n=6$). Significant difference between control and Mn-treated rats (* $p < 0.05$) by one-way ANOVA followed by Bonferroni's multiple comparison tests.

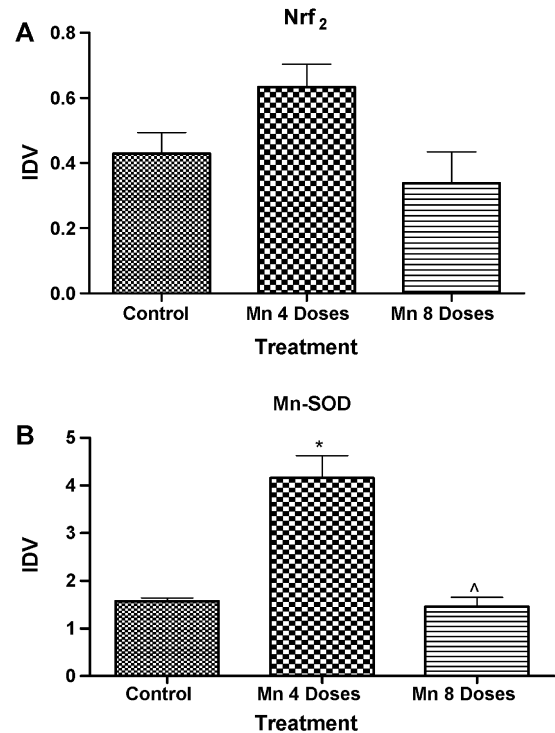


Fig. 8. (A) Nrf₂ protein levels in brain of rats intra-peritoneally injected with 4 or 8 doses of MnCl₂ and a control group. (B) Mn-SOD levels in brain of rats similarly exposed. Data are expressed as the mean \pm SEM ($n=6$). IOD, integrated optical density. Significant difference between control and Mn-treated rats (* $p < 0.05$) by one-way ANOVA followed by Bonferroni's multiple comparison tests. $\wedge p < 0.05$, significantly different from 4 Mn doses by one-way ANOVA followed by Bonferroni's multiple comparison tests. Representative examples of gels used to measure the activities of antioxidant enzymes. The Mn-SOD and Nrf₂ gels were analyzed as described in Section 2.

4. Discussion

Although Mn is an essential element which is present at low concentrations in food, water and air, exposure to excessive levels of this metal may lead to elevated brain Mn accumulation and adverse neurological effects (Crossgrove and Zheng, 2004; Aschner et al., 2005; Lu et al., 2005). The neurotoxic events of Mn are mediated, at least in part, via interference with central cholinergic systems. Finkelstein et al. (2007) hypothesized that the Parkinson's-like movement disorder inherent to excessive exposure to Mn results from imbalance between the DAergic and cholinergic systems of the basal ganglia. The delicate functional neurochemical balance is disrupted upon damage to both dopamine (DA) and cholinergic systems, caused either directly or indirectly via GABAergic and glutamatergic interneurons. While most research on Mn toxicity has focused on perturbations in the DAergic and GABAergic system, evidence exists in support of a direct effect of the DAergic system on striatal cholinergic function. Striatal cholinergic function is inhibited by nigrostriatal activity, since DAergic receptor blockers enhance ACh release and lower striatal concentrations of DA (Stadler et al., 1973). Previous *in vivo* studies show that Mn alters the activity of enzymes involved in cholinergic transmission, such as AChE (Deskin et al., 1981; Lai et al., 1984; Liapi et al., 2008) and choline acetyltransferase (Lai et al., 1981, 1984; Martinez and Bonilla, 1981). AChE is the degradative enzyme of ACh and thereby responsible for the termination of cholinergic response in muscarinic and nicotinic brain ACh receptors (Milatovic et al., 2006). Although none of the studies in which the activity of AChE was measured in brain of rats exposed acutely or chronically to Mn developed hypercholinergic signs, it was demonstrated that AChE activity is modulated by Mn exposure (see below).

Exposure to excessive levels of Mn may cause a variety of neurotoxic effects that involve (among others) alterations in biomarkers of oxidative stress (Taylor et al., 2006; Erikson et al., 2007). Liapi et al. (2008) investigated the effects of short-term exposure to Mn on the adult rat brain antioxidant status and the activities of AChE, noting that AChE activity was significantly increased in the Mn exposed group. Herein, we have used an *in vivo* model to test the hypothesis that Mn's effect on AChE activity contributes to oxidative stress injury in the rat's brain by measuring F₄-NPs and F₂-IsoPs levels. Our results show that after 4 or 8 dose Mn treatment, brain AChE activity significantly decreased ($p < 0.05$) to $60 \pm 16\%$ and $55 \pm 13\%$ of control levels, respectively (Fig. 4).

The results are contradictory to those of Liapi et al. (2008) and in agreement with Deskin et al. (1981), Lai et al. (1981, 1982), Martinez and Bonilla (1981) and Sitaramayya et al. (1974). Deskin et al. (1981) reported that AChE activity was significantly reduced in the striatum of rats treated with MnCl₂ (20 µg/g bodyweight/day), administered by oral gavage from birth to day 24 *post-partum*. Sitaramayya et al. (1974) reported a significant decrease of AChE activity in the cerebral cortex, cerebellum and the remaining cerebral tissue in rats treated with MnCl₂ inoculated intraperitoneally for 120 days. In contrast, Liapi et al. (2008) reported a significant increase of the activity of AChE in rats' whole brains treated daily with MnCl₂ 50 mg/kg for 7 days by the i.p. route. In the study by Lai et al. (1982), the distribution of AChE activity was assessed in various brain regions in the immediate postnatal period and found that the AChE activity was highest in the striatum, lowest in the cerebellum.

While none of the studies in which the activity of AChE was measured in the brain of rats exposed acutely or chronically to Mn developed hypercholinergic signs, there are several reports of seizures in patients receiving parenteral nutrition with hypermanganesemia (Fitzgerald et al., 1999; Komaki et al., 1999; Hsieh et al., 2007). Although cases of convulsions associated with Mn intoxication are a rare presentation of Mn intoxication, there is no satisfactory explanation for these symptoms. Seizures are clinical signs associated with hyperstimulation by ACh (Harris, 2006) and both seizures and convulsions are considered typical sequelae of systemic application of sublethal doses of AChE inhibitors (Milatovic et al., 2006).

Exposure to excessive Mn levels is associated with motor disturbances (Calne et al., 1994; Pal et al., 1999). Behavioral analysis on spontaneous activity of both ambulation and rearing showed a significant decrease ($p < 0.001$) following Mn administration vs. the control group (Fig. 2A and B, respectively). This is consistent with previous studies where rats exposed to MnCl₂ showed a decrease in spontaneous activity in open-field (Bonilla, 1984; Nachtman et al., 1986; Salehi et al., 2003; Torrente et al., 2005; Tapin et al., 2006; Vezér et al., 2007; Marreilha dos Santos et al., 2011). Our results may be explained by the accumulation of Mn in the striatum, which is a region involved in motor coordination that is under neuronal regulation of DA, ACh, GABA and glutamate (Moore and Wuerthele, 1979; Carlsson and Carlsson, 1990). Chronic exposure to Mn can lead to a movement disorder due to preferential Mn accumulation in the globus pallidus and other basal ganglia nuclei (Fitsanakis et al., 2008). Consistent with previous studies (Gianutsos et al., 1985), the Mn exposed groups showed a significant increase ($p < 0.05$) in brain Mn concentrations (Fig. 3) compared with the control group. Studies both in primates and rodents suggest that homeostatic control of brain Mn concentrations is tight, since even in conditions of high exposures, levels of brain Mn increase only several fold (3–4) (Erikson et al., 2004).

Bodyweight was measured several times along the experiment axis, establishing a trend for diminished weight-gain in the Mn-treated group as compared to the control group (Fig. 1). Similar results have been described in other studies (Torrente et al., 2005;

Vezér et al., 2007). This suggests that it is unlikely that malnutrition plays a significant role in the genesis of the neurochemical changes observed in this rat model.

F₂-IsoPs are a group of AA-derived prostanoid isomers generated from the free radical-initiated peroxidation of AA independent of COX (Musiek and Morrow, 2005; Milatovic and Aschner, 2009; Roberts and Milne, 2009). Analogously, auto-oxidation of the DHA generates F₄-NPs (Musiek et al., 2005). Analysis of cerebral biomarkers of oxidative damage showed Mn induced dose-dependent increase in brain F₂-IsoPs levels (Fig. 5). These results corroborate a recent study in mice (Milatovic et al., 2009) that showed a dose- and time-dependent increase in F₂-IsoPs levels in the brain after MnCl₂ exposure. Nevertheless, a dose-dependent increase in Mn-induced F₄-NPs generation was absent. Fig. 6 shows that only the 4 dose Mn treated rats caused a significant increase of F₄-NPs concentration, 24 h following the last injection. Our results show that F₄-NPs are earlier biomarkers of Mn-induced oxidative injury than F₂-IsoPs. This may be related to the fact that DHA is more susceptible to peroxidation than AA, due to its higher degree of unsaturation (six double bonds vs. four, respectively) (Greco and Minghetti, 2004). In the brain, DHA is preferentially distributed in gray matter, where it accounts for about one third of the total fatty acids of neuron phospholipids (Greco and Minghetti, 2004). AA is relatively evenly distributed in brain with similar concentrations in gray and white matter, as well as within glia and neurons. Thus, F₂-IsoPs quantification is a general reflection of oxidative damage in the whole brain, while F₄-NeuroPs in particular to gray matter (Reich et al., 2000, 2001). Accordingly, the quantification of NPs may provide a more selective and sensitive measure of oxidant injury than F₂-IsoPs. Nevertheless, since F₄-NPs levels didn't increase in a dose-dependent manner, likely due to its degradation and/or elimination from the brain, consideration should be given to the suitability of this biomarker in evaluating neuronal damage when used as a single biomarker of oxidative stress.

As an additional indicator of Mn neurotoxicity, we have evaluated the neuroinflammatory response by measuring PGE₂ levels. We observed a significant increase in PGE₂ levels in rat brain only after 8 doses with MnCl₂ (Fig. 7). These data corroborates a previous *in vivo* study (Milatovic et al., 2009) showing that MnCl₂ exposure caused a time- and dose-dependent increase in PGE₂ levels in mice brains. ROS generation is also associated with inflammatory responses and release of inflammatory mediators, including prostaglandins (Wang et al., 2004; Milatovic et al., 2009). These findings suggest that biomarkers of neuroinflammation may not be as sensitive as the biomarkers of oxidative stress in evaluating brain injury mediated by Mn.

Free radicals generated by oxidative stress cause damage that can contribute to numerous chronic diseases. Mammalian cells respond to this damage by increased transcription of cytoprotective phase II genes, which are regulated by Nrf2 (Kang et al., 2010). There are multiple genes regulated by Nrf2: phase II detoxification enzymes, as well as antioxidant proteins including glutathione-S-transferases, NAD(P)H:quinone oxidoreductase-1, γ-glutamylcysteine synthase, ferritin, and heme oxygenase-1 (Chen and Kunsch, 2004). Under stressful conditions, Nrf2 translocates into the nucleus and binds to the antioxidant response element (ARE), activating defensive gene expression (Kaspar and Jaiswal, 2011). The western blot analysis showed a significant increase of Mn-SOD protein expression and increased tendency towards increased Nrf2 expression after 4 Mn doses, reflecting an adaptive response to increased oxidative stress (Fig. 8A and B). Notably, we observed a decrease in cellular defense mechanisms associated with oxidative stress upon 8 Mn doses (demonstrating protein expression levels analogous to controls). Morello et al. (2007) also observed a reduction in the expression of Mn-SOD after chronic Mn treatment. Interestingly, a reduction in mitochondrial

Mn-SOD has been reported to exacerbate glutamate toxicity in cultured mouse cortical neurons (Li et al., 1998). Changes in Mn-containing proteins have been observed in many neurodegenerative diseases, including Alzheimer's disease, amyotrophic lateral sclerosis and Parkinsonian-like syndrome (Yin et al., 2008). Mn-SOD is critical in preventing or limiting apoptosis and necrosis resulting from cellular damage caused by ROS (Frankel et al., 2000; Raha and Robinson, 2000; Kitazawa et al., 2002). In addition, Mn-SOD, high intra-cellular catalase and glutathione levels offsets the apoptosis triggered by ROS (Simon et al., 2000).

The decrease in both proteins' expression, approaching control levels after 8 doses Mn treatment suggests this effect may predispose the cells to Mn-induced oxidative stress, which can compromise cell survival. Fujimura et al. (1999) found that Mn-SOD blocks cytosolic release of mitochondrial cytochrome c after ischemia, a critical step in the apoptotic process. As the previous study shows that Mn-SOD expression is a critical protein in the prevention of apoptosis process, at high levels of exposure to Mn, the cell is prone to apoptosis.

Consistent with the analogous Mn brain levels in animals exposed to 4 or 8 Mn injections (probably owing to saturation of transport mechanisms), AChE activity was also indistinguishable between the two groups. The F₂-IsoPs and PGE₂ are late biomarkers of the neurotoxic effect of Mn, since the anti-oxidant defenses (Mn-SOD and Nrf2) are challenged earlier in the time course to prevent the deleterious effects to the brain. Thus, it is not surprising that we found a poor correlation between F₂-IsoPs and AChE activity levels ($r^2 = 0.642$ only for the 8 doses). The effect of Mn on AChE activity contributes to the oxidative stress, consistent with the increase in SOD-Mn levels.

The present study shows that AChE inhibition in rats' brain, may lead to oxidative events and eventually disrupt the balance between ACh and DA in the cholinergic-dopaminergic system (Finkelstein et al., 2007). ACh is thought to act directly on DAergic terminals, possibly via collaterals, because cholinergically-induced release of DA occurs in the presence of tetrodotoxin, a neurotoxin that prevents the generation of action potentials (Moore and Wuerthele, 1979). ACh releases DA from isolated perfused striata (Besson et al., 1969) from striatal slices and from striata *in vivo* (Giorgiueff et al., 1976), while intravenous injections of atropine increase striatal concentrations of homovanillic acid (Moore and Wuerthele, 1979). Since Mn's effects are region-specific, future studies should be addressed on DA and ACh balance in various brain areas in order to clarify the neurochemical/pathophysiological mechanisms associated with its neurotoxicity.

In summary, the present study determined whether Mn effects on AChE activity play a role in oxidative stress and neurotoxicity. Our results indicate that Mn inhibits AChE concomitant with significant changes in the levels of biomarkers of oxidative stress (F₂-IsoPs and F₄-NPs). We posit that AChE is a target of Mn in the central nervous system that may trigger or contribute to the development of oxidative stress. These results suggest that AChE activity may serve as an early biomarker of Mn neurotoxicity and risk assessment inherent to populations exposed to this metal. AChE is not a marker of cholinergic function, since it is not a specific enzyme of the central cholinergic neurons. AChE is localized also within the nigrostriatal DAergic neurons (Butcher, 1977). Therefore, the utility of AChE activity as an early biomarker of Mn neurotoxicity should be further addressed in biomonitoring Mn neurotoxicity.

Conflict of interest statement

None of the authors has any financial interest or conflict of interest related to the manuscript.

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